

# Evidence of a 2<sup>nd</sup>-Order Peierls-Driven Metal-Insulator Transition in Crystalline NbO<sub>2</sub>

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The metal-insulator transition of NbO<sub>2</sub> is thought to be important for the functioning of recent niobium oxide-based memristor devices, and is often described as a Mott transition in these contexts. However, the actual transition mechanism remains unclear, as current devices actually employ electroformed NbO<sub>x</sub> that may be inherently different to crystalline NbO<sub>2</sub>. We report on our synchrotron x-ray spectroscopy and density functional theory study of crystalline, epitaxial NbO<sub>2</sub> thin films grown by Pulsed Laser Deposition and Molecular Beam Epitaxy across the metal-insulator transition at  $\sim 810$  °C. The observed spectral changes reveal a second-order Peierls transition driven by a weakening of Nb dimerization without significant electron correlations, further supported by our density functional theory modeling. Our findings indicate that employing crystalline NbO<sub>2</sub> as an active layer in memristor devices may facilitate analog control of the resistivity whereby Joule-heating can modulate Nb-Nb dimer distance and consequently control the opening of a pseudogap.

## I. INTRODUCTION

Niobium oxide-based memristor devices have demonstrated various switching behaviors making them attractive as building blocks for neuromorphic circuits.[1–4] In the majority of reports to date, disordered Nb<sub>2</sub>O<sub>5</sub> is deposited between metal electrodes and then activated using an electroforming process to reduce regions of the material to NbO<sub>x</sub>. The switching mechanism of the activated NbO<sub>x</sub> layer has then been explained in the literature in terms of both a Poole-Frenkel mechanism[5] and a Mott transition associated with NbO<sub>2</sub> nanostructures.[3, 6, 7] Without a good understanding of the true switching origin, it is difficult to further improve these devices and potentially circumvent electroforming with its associated scaling issues. It is therefore critically important to investigate the fundamental physics of the NbO<sub>x</sub> materials used within these devices and distinguish between the different proposed mechanisms. For example, recent studies of amorphous Nb<sub>2</sub>O<sub>5</sub>-based devices have revealed a Poole-Frenkel conduction mechanism that enables a memdiode behavior

without utilizing any electroforming.[8]

Recent advances in the quality of pure-phase, crystalline NbO<sub>2</sub> films[9] could enable device architectures that directly utilize the metal-insulator transition (MIT), assumed by many to be critical to the NbO<sub>x</sub> memristance. However, the nature of the NbO<sub>2</sub> MIT remains unclear and is often invoked as a Mott transition purely by comparison to VO<sub>2</sub>, wherein a cooperative Mott-Peierls transition occurs.[10–13] Density functional theory (DFT) calculations of crystalline NbO<sub>2</sub> polymorphs have instead supported a Peierls mechanism, since the insulating body centered tetragonal (BCT) and metallic rutile phases can both be reproduced without requiring electron-correlation effects to open and close a band gap.[14, 15] This has been further supported by recent DFT combined with the slave spin method (DFT+SS).[16] DFT+SS is capable of describing a strain-induced orbital selective Mott transition observed in VO<sub>2</sub>, which is seemingly not possible in NbO<sub>2</sub>. However, any attempts at direct comparison between these calculations and experimental x-ray spectroscopy results have suffered due to material issues associated with the surface-sensitive techniques employed[17] and the high MIT temperature ( $T_c = 1083$  K) of NbO<sub>2</sub>. [18]

Here, we report on our temperature-dependent synchrotron spectroscopy studies of crystalline NbO<sub>2</sub> films

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probing the true nature of the MIT. We employ hard x-ray photoelectron spectroscopy (HAXPES) and fluorescent yield x-ray absorption spectroscopy (XAS) to circumvent surface issues and fingerprint the bulk insulating and metallic phases of  $\text{NbO}_2$ . DFT calculations and many body simulations of the valence and conduction band spectra, respectively, are compared to the experimental spectra and reinforce that the band gap is formed mainly by Nb-Nb dimer formation. Variable photon energy and temperature-dependent soft x-ray photoelectron spectroscopy (XPS) are then used to characterize the film surface and distinguish between the surface and bulk band evolution occurring across the MIT. Finally, EXAFS fitting is used to confirm the temperature response of Nb dimers within the bulk insulating phase.

We report on the high temperature stability of  $\text{NbO}_2$ , which displays little to no bulk oxygen loss even after heating to 1600 °C. And we confirm the existence of an intrinsic and atomically-thin  $\text{Nb}^{5+}$  surface reconstruction that does not participate in the bulk dimerization. Our analysis reveals that bulk  $\text{NbO}_2$  undergoes a gradual temperature-dependent evolution of states at the Fermi level, which can be modeled in terms of a pseudogap that gradually opens with decreasing Nb-Nb dimer distance (i.e., a second-order Peierls-driven MIT). These results indicate that Mott physics does not play a significant role in the transition of crystalline  $\text{NbO}_2$ , with only weak contributions possible from any electron correlations due to the wide Nb 4d orbital bandwidth. Our work clarifies the true nature of the MIT in crystalline  $\text{NbO}_2$  as Peierls-driven and reveals the importance of Nb dimer distance to the electronic conductivity.

## II. METHODS

Crystalline  $\text{NbO}_2$  films were grown using both pulsed laser deposition (PLD) and molecular beam epitaxy (MBE). Approximately 400 nm thick, (440)-oriented  $\text{NbO}_2$  films were grown epitaxially on 430  $\mu\text{m}$  thick,  $10 \times 10$  mm, single crystalline, c-plane (006)-oriented  $\text{Al}_2\text{O}_3$  substrates using a Varian Gen-II style MBE system at the Georgia Institute of Technology. Prior to deposition, substrates were cleaned with acetone, methanol, and isopropyl alcohol followed by a 4:1  $\text{H}_2\text{SO}_4\text{:H}_2\text{O}_2$  wash at 120 °C. Substrates were then loaded and degassed in a separate chamber, with care taken during loading to minimize sample contact with the surrounding metal substrate holder which will conduct thermal energy away from the sample creating thermal non-uniformity. Thin films were then grown under a 4 SCCM  $\text{O}_2$  flow at high temperatures (950 - 1000 °C). It should be noted that higher temperatures than typically used in MBE were necessary to achieve high quality  $\text{NbO}_2$  films. Additional growth details have been previously reported.[9]

In addition, (100)-oriented  $\text{NbO}_2$  films were grown on single crystalline, (110)-oriented  $\text{TiO}_2$  substrates using a PLD system at the Leibniz-Institut für Kristallzüchtung.

Prior to deposition, substrates were annealed for 2 hours at 1000 °C in an oxygen atmosphere. The thin films were subsequently deposited via PLD using a KrF excimer laser operated at 1 Hz to ablate a  $\text{NbO}_2$  target with a 60 mm substrate to target distance. Depositions were carried out under a 0.1 mbar Ar atmosphere with a 650 °C substrate temperature. Depositions were monitored using in-situ RHEED. The fluency was 2.1/2.5 J/cm<sup>2</sup> and 3000/20000 pulses were used for each deposition resulting in films with a thickness of approximately 18 nm/111 nm, as determined from the thickness fringes around the  $\text{NbO}_2$  (400) diffraction peak.

Hard x-ray photoelectron spectroscopy (HAXPES) was performed with a 5932 eV photon energy at the I09 beamline of the Diamond Light Source (DLS) in Oxfordshire, UK. Spectra were energy-resolved and measured using a VG Scienta EW4000 high-energy electron-energy analyzer with a  $\pm 30^\circ$  acceptance angle. The photon beam was monochromated using a Si(111) double-crystal monochromator followed by a channel cut Si(004) crystal, resulting in an energy resolution of < 250 meV.[19–21]

Variable photon energy x-ray photoelectron spectroscopy (XPS) was performed at beamline 29-ID of the Advanced Photon Source (APS) at Argonne National Laboratory in Lemont, IL. Temperature-dependent x-ray photoelectron spectroscopy (T-XPS) was performed at the Electron Spectro-Microscopy (ESM) beamline 21-ID of the National Synchrotron Light Source-II at Brookhaven National Laboratory in Upton, NY.

O K-edge x-ray absorption spectroscopy (XAS) was performed at beamline 8 of the Advanced Light Source (ALS) in Berkeley, CA. Spectra were taken in both total electron yield (TEY) and total fluorescence mode (TFY), with an effective resolution of  $\sim 200$  meV. Spectra were normalized to the current from an Au-coated reference mesh in the path of the incident photon beam. Energy calibration was performed using the first- and second-order diffraction Ti  $L_{3,2}$ -edge and O K-edge absorption features of a rutile  $\text{TiO}_2$  reference.

Temperature-dependent Nb K-edge extended x-ray absorption fine structure (EXAFS) measurements were performed on commercially produced  $\text{NbO}_2$  powder (CAS Number 12034-59-2) at beamline 20-BM of the APS. Measurements were performed only up to 400 °C, as oxidation to  $\text{Nb}^{5+}$  was found to occur when annealing at atmospheric pressures, even when using an inert gas flow to reduce oxygen exposure. Energy calibration was performed by energy shifting the maximum of the first peak in the derivative spectrum of a pure Nb reference foil to the Nb K-edge reference energy (18986 eV). Data processing was carried out using the ATHENA software from the IFEFFIT package.[22, 23] The undulatory region of the normalized spectrum was first converted into k-space. The resultant  $\chi(k)$  was then  $k^2$ -weighted to enhance higher  $k$  in the Fourier transformation of  $k^2\chi(k)$ , thus enabling model peak fitting. To resolve the Nb-Nb dimer peak and inter-atomic length precisely, a theoretical model spectrum was generated

via ab-initio calculations with relevant structural data using the FEFF8.2 code[24] and then least-square fit to the data using ARTEMIS.

Density functional theory (DFT) calculations were performed within the WIEN2k[25] ecosystem employing full potential linear augmented plane waves plus local orbitals (FP-LAPW+lo) and the generalized gradient approximation of Perdew-Burke-Erzenhof (PBE).[26] The total energy was converged to 0.1 meV on a 1000 k-point grid with an RKmax of 6.5. Ionic relaxations were performed with VASP[27, 28] with a cutoff energy of 750 eV on an  $8 \times 8 \times 12$  and  $8 \times 8 \times 8$  k-mesh for rutile and BCT phases. Many body x-ray absorption simulations were performed to using a constrained-occupancy delta-self-consistent-field approach rectified by explicitly calculating many-electron transition amplitudes, as described by Prendergast et al.[29]

### III. RESULTS

$\text{NbO}_2$  is known to undergo a transition between an insulating BCT phase containing Nb dimers and a more symmetric, metallic rutile phase around  $\sim 810^\circ\text{C}$ , both depicted in Figure 1 (a). HAXPES valence band spectra and fluorescent yield XAS O K-edge spectra, shown in Figure 1 (b) and (c) respectively, reflect the density of occupied valence states and partial density of unoccupied O 2p conduction states in the low and high temperature phases of crystalline  $\text{NbO}_2/\text{TiO}_2$ . Corresponding Nb 3d core-level HAXPES confirms a  $\text{Nb}^{4+}$  oxidation state in the bulk of the film (refer to supplemental information). The increased probing depth afforded by these high photon energy methods reduces the spectral contamination from  $\text{Nb}^{5+}$  surface components observed in other studies,[17] as discussed further below.

The valence band spectra display the occupied Nb 4d  $t_{2g}$  state clearly separated from the main O 2p-derived valence band and centered  $\sim 1$  eV below the Fermi level in the insulating phase, in agreement with DFT calculations. Consistent with previous reports of weak electron correlation effects, the band theory calculations are sufficient to accurately reproduce a band opening without the need for +U corrections, although they underestimate the band gap magnitude.[14–17] Unfortunately, the small band gap of the low temperature BCT phase means the O K-edge XAS cannot be compared to the unoccupied O 2p partial density of states calculated using DFT. To account for the many-body dynamic excitations occurring due to  $\text{NbO}_2$  being a near-metallic oxide, we employed a self-consistent-field approach by Prendergast et al.[29] The simulated absorption for the dimerized BCT phase is found to agree well with the low temperature experimental data. Taken together, these results are consistent with the insulating phase arising from mainly Nb dimerization (Peierls) rather than electron correlation (Mott).

To investigate the electronic evolution of  $\text{NbO}_2$  across the MIT, surface sensitive XPS was performed on

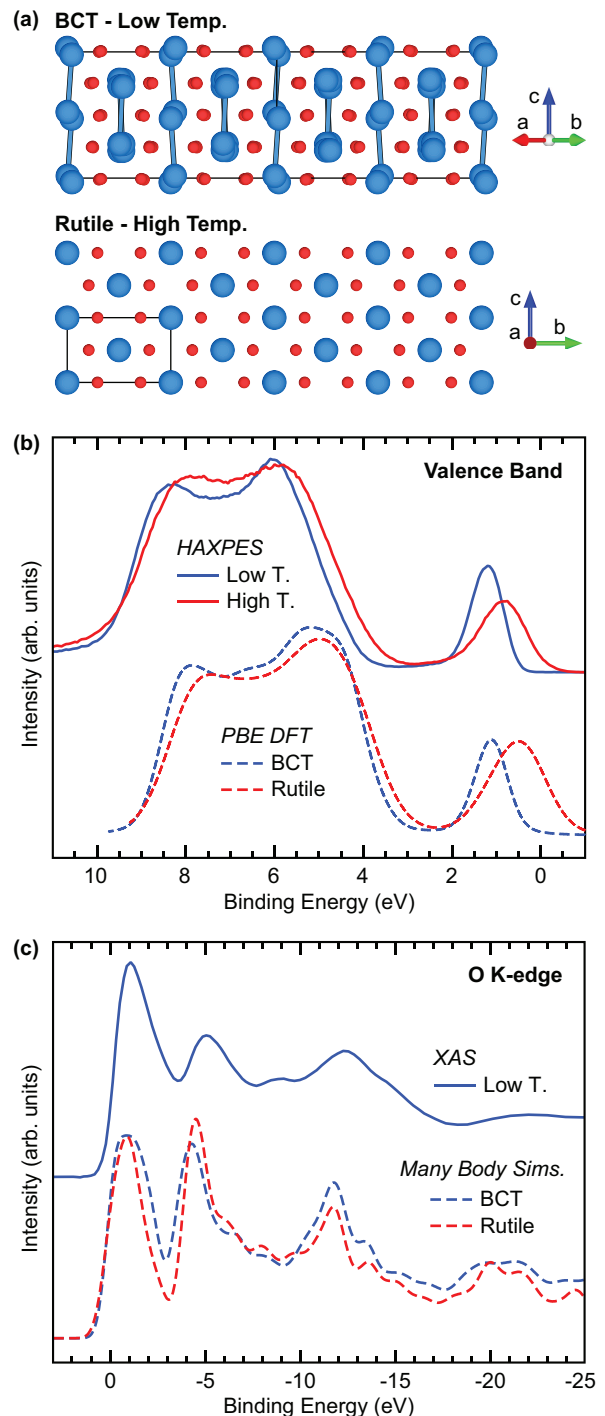


FIG. 1. (a) Low temperature BCT and high temperature rutile crystal structures of  $\text{NbO}_2$ . (b) Valence band and (c) oxygen k-edge spectra, both measured and calculated, for the low temperature BCT (blue) and high temperature rutile (red) phases of a crystalline  $\text{NbO}_2/\text{TiO}_2$  thin film.

$\text{NbO}_2/\text{Al}_2\text{O}_3$  at several soft x-ray synchrotron beamlines. To remove surface contaminants and reduce any over-oxidation of the films arising from atmospheric exposure that might complicate results from these surface sensitive

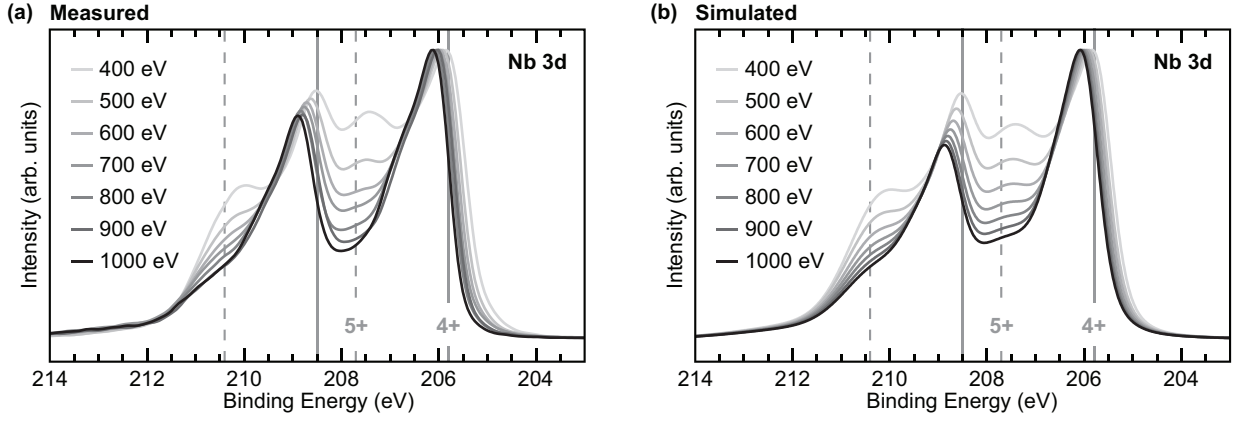


FIG. 2. (a) Variable photon energy XPS of the Nb 3d region for  $\text{NbO}_2/\text{Al}_2\text{O}_3$  at room temperature, and (b) simulated spectra assuming a 0.9 nm  $\text{Nb}_2\text{O}_5$  ( $\text{Nb}^{5+}$ ) surface layer on top of the  $\text{NbO}_2$  ( $\text{Nb}^{4+}$ ) bulk. Approximate  $\text{Nb}^{4+}$  and  $\text{Nb}^{5+}$  spin-split peak energies are identified.

techniques, clean and well-ordered sample surfaces were achieved by annealing the films in vacuo at high temperatures near and above the MIT. Sapphire substrates were used here as the  $\text{TiO}_2$  substrate used for the previous HAXPES experiments proved to be too unstable at high temperatures, resulting in oxygen migration from the substrate and into the  $\text{NbO}_2$  film. After preparation, the sample surfaces were thoroughly characterized before investigating the MIT behavior to ensure that any results are properly attributed to either bulk or surface effects.

Figure 2 (a) displays non-destructive depth profiling of the  $\text{NbO}_2/\text{Al}_2\text{O}_3$  performed using Nb 3d core-level spectra measured at multiple incident x-ray photon energies. The spectra reveal both  $\text{Nb}^{4+}$  and  $\text{Nb}^{5+}$  oxidation states existing within the film. However, the lower the incident photon energy (i.e., the more surface sensitive the measurement), the more enhanced the  $\text{Nb}^{5+}$  peaks become relative to the  $\text{Nb}^{4+}$ , indicating the  $\text{Nb}^{5+}$  is restricted to the very topmost surface layer. Shown in Figure 2 (b), simulated XPS spectra were created by modeling the film as a two layer system with a  $\text{Nb}_2\text{O}_5$  ( $\text{Nb}^{5+}$ ) surface layer on top of an  $\text{NbO}_2$  ( $\text{Nb}^{4+}$ ) bulk film (refer to supplemental information). A  $\text{Nb}_2\text{O}_5$  surface layer thickness of 0.9 nm is found to result in good agreement with the experimentally observed trend. This confirms that the  $\text{Nb}^{5+}$  is restricted to only the topmost atomic layers ( $\sim 1$  nm) of the  $\text{NbO}_2$  film.

Further supporting this conclusion, only a slight reduction in the amount of surface  $\text{Nb}^{5+}$  was observed after annealing above the MIT temperature for the first time (refer to supplemental information). And no amount of further annealing up to  $1600^\circ$  removed the remaining  $\text{Nb}^{5+}$  surface layer, nor did it reduce the bulk  $\text{Nb}^{4+}$ , as confirmed by HAXPES (refer to supplemental information). The robust nature of  $\text{NbO}_2$  regarding oxygen loss at high temperatures is note-worthy given the potential role of Joule-heating in the switching of  $\text{NbO}_x$ -based devices. This data suggests that our  $\text{NbO}_2/\text{Al}_2\text{O}_3$  film has a strong proclivity towards forming an inert  $\text{Nb}^{5+}$  sur-

face, likely the result of a preferred surface termination rather than an over-oxidization of the film. This inert surface must be accounted for when using highly surface sensitive techniques like XPS, LEED, or LEEM/PEEM to investigate the MIT behavior.

Figure 3 shows temperature-dependent XPS of the O 1s, Nb 3d, and valence band regions of the  $\text{NbO}_2/\text{Al}_2\text{O}_3$ . The spectra were taken in descending order (starting at  $1000^\circ\text{C}$  and then cooling down in steps until room temperature) in order to rule out surface composition/oxidation changes due to in vacuo heating as the cause of increased metallicity at high temperatures. In addition, these temperature-dependent measurements were repeated multiple times to fully ensure the spectral changes reported herein are reversible and not due to irreversible sample damage such as oxygen loss (refer to supplemental information).

The transition between metallic and insulating states is readily apparent from changes near the Fermi level. Spectral weight is found to shift towards the Fermi level with increasing temperature, causing the eventual closure of the band gap at the MIT temperature. This transition is gradual, occurring over a wide range of several hundred degrees, rather than abruptly as would be expected for a first-order transition like in  $\text{VO}_2$ .<sup>[30]</sup> This observation is consistent with previous temperature-dependent conductivity studies of  $\text{NbO}_2$  crystals,<sup>[18]</sup> where an almost 9 orders of magnitude change in conductivity was found to gradually occur over a wide temperature range below the official MIT temperature. It must be noted that while the films display both  $\text{Nb}^{5+}$  'surface' and  $\text{NbO}_2$  'bulk' signal contributions due to the surface sensitivity of the XPS, the  $\text{Nb}^{5+}$  surface does not contribute any spectral weight to the occupied Nb 4d states at the Fermi level responsible for the change in metallicity.

As the  $\text{NbO}_2$  is heated towards the transition temperature, two different effects are observed in the XPS spectra which must be disentangled to understand the MIT. There is a symmetric thermal broadening affect-

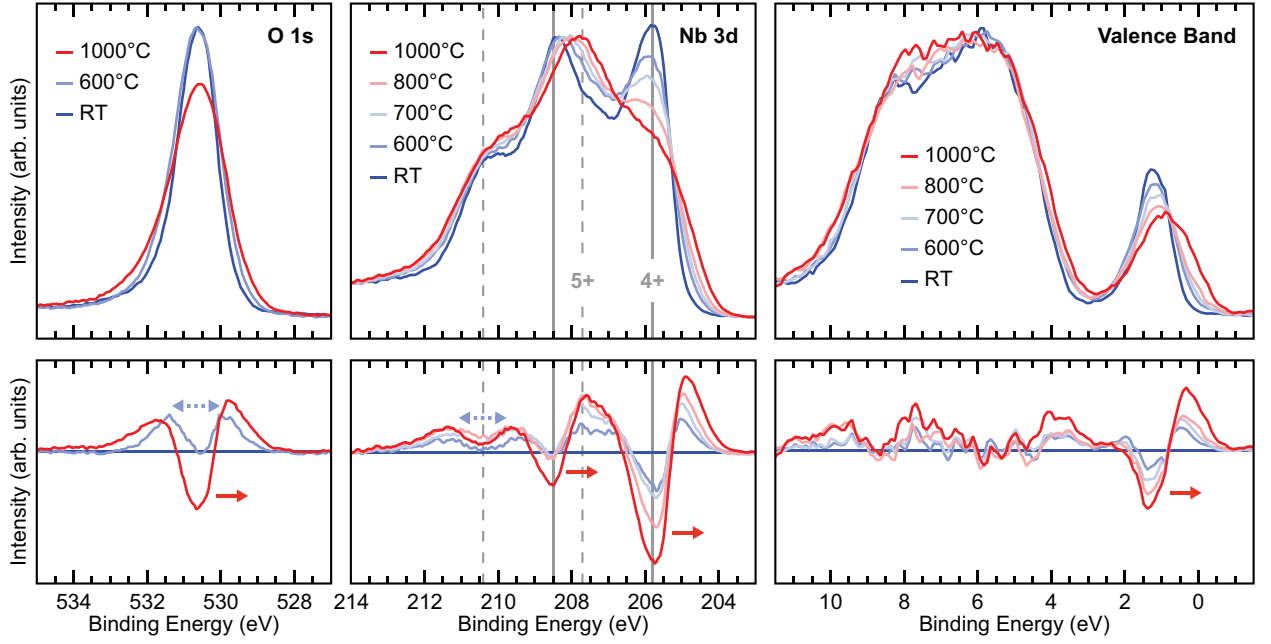


FIG. 3. (top) Temperature-dependent XPS of the O 1s region ( $h\nu = 700$  eV), and the Nb 3d and valence band regions ( $h\nu = 450$  eV) of  $\text{NbO}_2/\text{Al}_2\text{O}_3$ . Nb 3d peak positions for  $\text{Nb}^{5+}$  and  $\text{Nb}^{4+}$  charge states are denoted. (bottom) Difference spectra showing the spectral changes in the O 1s, Nb 3d, and valence band regions relative to the room temperature measurements. Red solid arrows indicate asymmetric broadening, while blue dashed arrows indicate symmetric broadening.

ing all spectral features and an asymmetric broadening to lower binding energies only affecting certain spectral features. These effects are more readily distinguished in the difference spectra plotted below the XPS, which use the room temperature XPS spectra as an internal reference. The symmetric broadening (blue dashed arrows) is most easily observed in the  $\text{Nb}^{5+}$  contributions of the Nb 3d region. Meanwhile, the asymmetric broadening (red solid arrows) is clearly observed in the  $\text{Nb}^{4+}$ -related states. The asymmetric broadening is consistent with a modulation of final state screening effects associated with the formation of states at the Fermi level (i.e., an increase in metallicity). The O 1s region further supports this, displaying only symmetric thermal broadening up to the MIT temperature, at which point an asymmetric broadening abruptly appears. Similar O 1s evolution is observed across the MIT of other metal oxides.[31] As mentioned previously, the transition is reversible with the  $\text{NbO}_2$  fully recovering to its initial state upon cooling.

Figure 4 (a) plots the DFT quasiparticle density of states near the Fermi level for  $\text{NbO}_2$  as the Nb dimer length varies between that of the fully dimerized BCT phase and the non-dimerized rutile phase. For the electronic structure simulations, the high and low temperature phases were both represented in a single ( $I4_1/a88$ ) body centered tetragonal space group.[17] Using a common lattice was necessary to interpolate the structural phases between the experimentally determined rutile and BCT endpoints. Connecting the two phases was then achieved by linearly interpolating the Nb-Nb bond lengths along the dimerization axis (c-axis).

As the Nb dimers form (Nb-Nb bond length decreases), there is a concomitant increase in electron localization. This gradually increasing localization results in the formation of a pseudogap that eventually opens into a full band gap, creating a fully insulating phase.[14, 15, 17] This is in contrast to the action of a Hubbard  $U$  correction, which does not gap the spectrum in the metallic phase, nor does it act to significantly change the electronic character in the insulating phase (refer to supplemental information). This shows that the number of occupied states at the chemical potential is highly sensitive to lattice deformations, and only weakly governed by electronic correlations. However, it is not just the form of the quasiparticle density of states that suggests a strong Peierls mechanism, but also the excellent agreement observed between the hard/soft x-ray spectral features and the calculations performed at the PBE level. These results indicate that electron-electron contributions to the ground state energy must be small compared to contributions from the lattice degrees of freedom.

Temperature-dependent EXAFS fitting results, shown in Figure 4 (b), also support the idea of a Nb dimer weakening with increased temperature. As temperature increases, the average Nb-Nb dimer length is found to increase in a manner that coincides with the predicted opening of a pseudogap, as calculated using both PBE DFT and the generalized solid-state nudged elastic band (G-SSNEB) method.[14] This experimental confirmation of a gradual structural transition reinforces the idea of a second-order mechanism relating to Nb dimerization. Combining this with the previously described confirma-



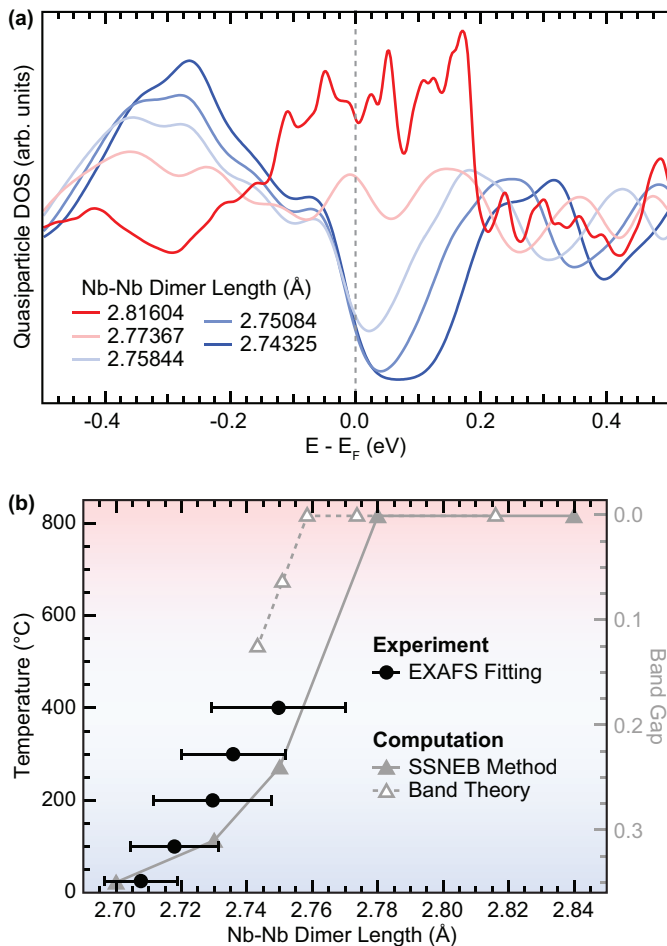


FIG. 4. (a) PBE DFT calculated density of states at the Fermi level for crystalline  $\text{NbO}_2$  with increasing Nb dimer distance. (b) Temperature-dependent EXAFS fitting results of the Nb dimer distance (left axis) compared to DFT and SSNEB[14] calculated bandgaps (right axis).

tion of a gradual electronic transition and the lack of any observed electron correlation effects, we conclude that the MIT behavior of  $\text{NbO}_2$  is best described in terms of a second-order, temperature-dependent Peierls transition, rather than a Mott transition.

#### IV. CONCLUSION

Crystalline  $\text{NbO}_2$  is conclusively found to undergo a second-order, temperature-dependent Peierls transition with little to no contribution from electron correlations. Our results indicate that as the temperature within crystalline  $\text{NbO}_2$  is increased by local Joule-heating, an insulator-to-metal phase transformation would likely occur due to a weakening of local Nb-Nb dimers, resulting

in memristor-like switching. Moreover, our results imply that the use of crystalline  $\text{NbO}_2$  as the active layer in next-generation memristive components has two major benefits. Firstly, an electroforming step would not be required to induce the MIT-based switching behavior, making it more appealing for industrial scale employment. Secondly, precise control of local Joule-heating is expected to enable analog modulation of the resistive state, due to the analog temperature-sensitivity of the Nb-Nb dimer length within the BCT phase. In addition, although strain-engineering epitaxial  $\text{NbO}_2$  films is not expected to increase electron correlation effects because of the Nb 4d bandwidth, it could potentially be employed as a method to tune Nb-Nb dimer distance for further control of resistive states.

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